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Electrical transport and magnetoresistance in La_{2/3}Ca_{1/3}MnO₃/BaZrO₃ composites

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1. Introduction

Much attention has been paid on perovskite oxides during the last decades because they have rich physical properties including superconductivity with high- $T_{\rm C}$, ferromagnetism with half metallicity, and ferroelectricity with large polarizations, etc. For manganite-based perovskite oxides of $La_{1-x}A_xMnO_3$ with an optimal doping content the colossal magnetoresistance (CMR) is observed when a high magnetic field is applied. The magnetoresistance effect is greatly influenced by the presence of grain boundaries in polycrystalline materials [1,2] and the lattice mismatch strain in heteroepitaxial films [3,4]. For potential applications in spintronics devices a large magnetoresistance effect at low applied magnetic field is necessary. A lot of efforts have been devoted to achieve low-field magnetoresistance in magnetic tunnel junctions (MTJs) with insulating barrier [5-12], in artificial bicrystal grain boundary junctions [13-18] and in manganite/insulating oxide composites [19-24]. In magnetic tunnel junctions the tunneling magnetoresistance (TMR) effect is principally dependent on the spin polarization of ferromagnetic electrodes and the spindependent tunneling process through insulating barrier. For the artificial bicrystal grain boundary junctions, isolated grain boundary effect on low field magnetoresistance can be investigated. Well-defined manganite/barrier interface and manganite grain boundary structure must be obtained in both cases above.

ABSTRACT

We report the results of electrical resistivity and magnetoresistance (MR) measurements carried out on La_{2/3}Ca_{1/3}MnO₃/BaZrO₃ composites prepared by solid state reaction method. The metal-to-insulator transition and the magnitude of magnetoresistance are significantly influenced by the content of BaZrO₃, suggesting the presence of ionic diffusion between La_{2/3}Ca_{1/3}MnO₃ and BaZrO₃, although the dependence of the electrical resistivity as a function of La_{2/3}Ca_{1/3}MnO₃ volume fraction can be roughly described by classical percolation conduction model. The conductive behavior above the metal-to-insulator transition temperature can be well explained by Mott's variable range hopping model.

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Usually conventional solid state reaction is employed to prepare manganite/insulating oxide composites, where the secondary insulating phase is assumed to be chemically immiscible with the manganite matrix. Enhanced spin-dependent scattering of conductive electrons at manganite/oxide phase boundary is expected to contribute to low-field magnetoresistance effect in the composites. In La_{2/3}Sr_{1/3}MnO₃/CeO₂ composite the magnetoresistance of 1.5% at 100 Oe was observed for the sample with a percolation threshold of $La_{2/3}Sr_{1/3}MnO_3$ volume fraction ($f_{LSMO} = 0.20$) [25], which is close to the critical volume fraction theoretically predicted for usual three-dimensional lattice types [26]. Similar experimental results for the manganite/insulating oxide composites were reported [21]. In the real systems, possible ionic diffusion between the phases, however, is important for both theoretical and experimental aspects for understanding behavior of conductivity for composite materials. We present detailed preparation, electrical transport and magnetoresistance studies of the La_{2/3}Ca_{1/3}MnO₃/BaZrO₃ composite. We observe that the effect of ionic interdiffusion between the La_{2/3}Ca_{1/3}MnO₃(LCMO) and BaZrO₃ secondary phase (BZO) should be taken into account for explanation of transport properties.

2. Experiment

Conventional solid state reaction in air was used to prepare La_{2/3}Ca_{1/3}MnO₃ bulks. These starting powders of La₂O₃, CaCO₃ and MnO₂ (99.5% purity) were mixed and heated at 1000 °C for 36 h in air with intermediate grinding. Cold-pressed pellets were then sintered at 1400 °C for 12 h with furnace cooling. For preparation of polycrystalline (x)LCMO:(1 – x)BZO samples with different La_{2/3}Ca_{1/3}MnO₃ volume fractions x, milled LCMO powder was thoroughly mixed and grounded with BaZrO₃ powder. The mixture was finally sintered at 1300–1400 °C for 24 h in air.

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Fig. 1. XRD pattern for (x) $La_{2/3}Ca_{1/3}MnO_3$: $(1 - x)BaZrO_3$ composites with different $La_{2/3}Ca_{1/3}MnO_3$ volume fraction x.

The phase composition and the crystalline quality of the La_{2/3}Ca_{1/3}MnO₃ and composites were analyzed employing X-ray diffraction (XRD) using Cu-K α radiation. The micrograph of samples was investigated by scanning electric microscopy (SEM). Resistivity dependent on temperature from 77 K to 300 K was measured by standard four-probe method in zero field and 0.6 T applied magnetic field to evaluate electrical and magnetoresistance properties for bar-shaped samples. The magnetoresistance (MR) is defined as MR=(R(0) - R(H))/R(0), where R(0) is the resistivity in zero magnetic field and R(H) is the resistivity in applied magnetic field of 0.6 T.

3. Results and discussion

X-ray powder diffraction result on La_{2/3}Ca_{1/3}MnO₃ in Fig. 1 demonstrates a pure perovskite phase within the experimental error. For (x) $La_{2/3}Ca_{1/3}MnO_3:(1-x)$ BaZrO₃ composites, no impurity phase peaks except for diffraction peaks of the La_{2/3}Ca_{1/3}MnO₃ and BaZrO₃ phase were detected. The relative intensities of diffraction peaks from the LCMO phase systematically change with the LCMO volume fraction. We notice that the (110) diffraction peak positions of the La_{2/3}Ca_{1/3}MnO₃ phase have a rather slight shift towards low-angle direction with increasing the BaZrO₃ content, whereas the (110) diffraction peak of the BaZrO₃phase moves towards high-angle direction. This suggests the presence of interdiffusion between Mn^{3+} and Zr^{4+} discussed later. It is actually difficult to get a complete chemical immiscibility between La2/3Ca1/3MnO3 and BaZrO3 if usual solid state reaction method is employed, although the ionic diffusion between the La_{2/3}Ca_{1/3}MnO₃ and BaZrO₃ phases does not lead to formation of impurity phases in the composites.

Fig. 2 gives SEM images of the (x)La_{2/3}Ca_{1/3}MnO₃:(1 – x)BaZrO₃ composites to study morphology evolution with introduction of BaZrO₃. Pure La_{2/3}Ca_{1/3}MnO₃ has a typical equiaxed microstructure with an average grain size of ~4 μ m. Upon addition of 10% BaZrO₃ the microstructure of the composite is apparently changed, where a lot of small grains are detectable. Further increasing the BaZrO₃ volume fraction obviously leads to formation of microstructure with homogenous fine grains with an average size of 0.4–0.5 μ m. For the *x* = 0.2 sample, however, sudden coarsening is controlled by the growth behavior of BaZrO₃ itself. We believe that in the (x)La_{2/3}Ca_{1/3}MnO₃:(1 – x)BaZrO₃ system the introduction of BaZrO₃ significantly suppresses the growth of La_{2/3}Ca_{1/3}MnO₃

grains due to blocking of movement of the grain boundaries. This kind of refined grain microstructure will affect the electrical transport property in these composites.

We measured temperature dependent resistivity curves in zero and 0.6 T applied magnetic field for a series of specimens with varying *x*. For x > 0.23 we observe an obvious metal-to-insulator transition. In Fig. 3 representative resistivity curves dependent on temperature for $(x)La_{2/3}Ca_{1/3}MnO_3:(1-x)BaZrO_3$ composites are plotted. The transition peak temperature is denoted as T_p . On the contrary, the samples with x < 0.23 do not display any resistive transition in the measured temperature and resistance range.

Fig. 4 indicates the transition peak temperature T_p with different LCMO volume fractions for the composites. With addition of a little amount of $BaZrO_3$ the T_p of the composite significantly drops down to 125 K for x = 0.90. Further increasing the BZO content gives rise to a slightly rising of T_p , but the T_p value is only around 180 K for $x \sim 0.40$. The T_p drop in the LCMO/BZO composite is much larger than that in other manganite/insulator composites [27,21]. This can be attributed to a possible variation of the manganite composition due to the existence of ionic interdiffusion between the phases. The reversible increase in T_p with decreasing the LCMO content for x < 0.90 implies that the ionic interdiffusion occurs mainly at the interface of the La_{2/3}Ca_{1/3}MnO₃ and BaZrO₃ phases. The magnetoresistance of the composite shows a reverse trend with the volume fraction of BaZrO₃, i.e. the MR increases with decreasing the *T*_p similar to the effects induced by element substitution [28] and lattice strain [4].

Fig. 5 shows the resistivity of the composites at 300 K as a function of the La_{2/3}Ca_{1/3}MnO₃ volume fraction. A sharp increase of five orders of magnitude in the resistivity of composites was detected at $x \sim 0.20$ in the measured range. The classical percolation law is usually used to evaluate the conductive process for the conductive regime. According the percolation law $\rho \propto (f - f_c)^{-t}$, [29] the percolation threshold f_c the resistivity exponent t fitted to the measured data are 0.195 and 1.9, which actually fall in the range of theoretical predications of three-dimensional continuum percolation models. The 0.195 percolation threshold basically agrees with the value found for $La_{2/3}Sr_{1/3}MnO_3/CeO_2$ composites ($f_{LSMO} = 0.20$) [21]. Because many factors, such as porosity, poor connectivity, grain size difference and composition variation, affect the percolation conductive behavior in conducting/insulating composites, we should point out that the fitting does not fully confirm the presence of pure percolative conductivity in the La_{2/3}Ca_{1/3}MnO₃/BaZrO₃ composites, where the apparent decrease of the transition peak temperature T_p with the introduction of BaZrO₃ is an indicator of composition variation in the La_{2/3}Ca_{1/3}MnO₃.

It has been found above that the electronic conduction at high T with the $La_{2/3}Ca_{1/3}MnO_3$ volume fraction follows the classical 3D continuum model for the LCMO/BZO composite. In order to explore the conductive mechanism above T_p we analyze the temperature dependent resistivity for the composites. For the pure La_{2/3}Ca_{1/3}MnO₃ sintered bulk the electronic conduction above $T_{\rm p}$ can be usually described by the thermal activation model $\rho = \rho_{\infty} \exp(E_0/\kappa T)$ [21]. For the LCMO/BZO composite it was not successful to fit the $\rho(T)$ data above T_p by the thermal activation model. The measured data in the relatively wide temperature range are well fitted by the variable range hopping (VRH) model $\rho = \rho_0 \exp[(T_0/T)^{1/4}]$ proposed by Mott [30]. Fig. 6 displays several typical results for f_{LCMO} = 0.23, 0.30, 0.70 and 0.90. We find that the slope obtained from the relationship between $\ln \rho$ and $T^{-1/4}$ as shown in Fig. 6 is almost identical for different x, suggesting that increasing the BZO content does not lead to a change of electronic conduction mechanism in the composite. This also indicated a very low and limited solution of BaZrO₃ in La_{2/3}Ca_{1/3}MnO₃. Based on consideration of the tendency of the T_p and MR as indicated in Fig. 4, we believe that the ionic diffusion, especially at the phase interface,



Fig. 2. SEM images of (x)La_{2/3}Ca_{1/3}MnO₃:(1 - x)BaZrO₃ sintered bulks.



Fig. 3. Temperature-dependent resistivity curves for (x)La_{2/3}Ca_{1/3}MnO₃:(1 – x)BaZrO₃ composites measured at zero field (solid lines) and 0.6 T (dotted lines).



Fig. 4. Metal-to-insulator transition temperature T_p (solid circles) and magnetoresistance MR (open circles) dependent on the La_{2/3}Ca_{1/3}MnO₃volume fraction x. The MR was measured at 0.6 T and given at near T_p , i.e. the maximal MR value.



Fig. 5. Electrical resistivity at 300 K as a function of the $La_{2/3}Ca_{1/3}MnO_3$ volume fraction x. The line corresponds to the fit by percolation law.



Fig. 6. $\ln \rho$ is plotted versus $T^{-1/4}$ for some representative $(x) \text{La}_{2/3} \text{Ca}_{1/3} \text{MnO}_3: (1-x) \text{BaZrO}_3 \text{ composites.}$ The lines are linear fits to the data by the formula $\rho = \rho_0 \exp[(T_0/T)^{1/4}]$ above T_p .

is responsible for the VRH-type conductivity in the LCMO/BZO composite. The ionic diffusion limited at the LCMO/BZO phase interface may give rise to random potential fluctuations, which localize the carriers in the LCMO.

Substitutive ionic diffusion in the LCMO/BZO composite occurs most probably between Mn³⁺ and Zr⁴⁺. The probability of the Mn^{3+}/Zr^{4+} interdiffusion should have a maximum at the phase interface. For the high f_{LCMO} composite the BaZrO₃ easily distributes homogenously at the grain boundaries of La_{2/3}Ca_{1/3}MnO₃, resulting in a heavy diffusion. The Mn³⁺ composition variation due to the ionic diffusion will be unfavorable to the double-exchange interaction, and the metal-to-insulator transition is thus suppressed. In the meantime, this kind of ionic interdiffusion can lead to a change of high-temperature conductivity mechanism like that of Mn-site doping effects. Although the limited ion substitution greatly has changed the metal-to-insulator transition temperature and corresponding high-temperature conductivity in the composites, the refined grain microstructure by addition of BaZrO₃ makes the system to keep three-dimensional continuum percolation behavior. It should be pointed out that a large low-field magnetoresistance of 53% in 0.6 T for the x = 0.90 composite was found, which is beneficial to practical applications for CMR materials.

4. Conclusion

We have prepared the La_{2/3}Ca_{1/3}MnO₃/BaZrO₃ composites by conventional solid state reaction method. The electrical and magnetoresistance properties of the composites have been investigated. Remarkable changes of metal-to-insulator transition temperature, low-field magnetoresistance and high-temperature resistivity behavior were observed, suggesting importance of the variation of the manganite composition due to ionic interdiffusion in the La_{2/3}Ca_{1/3}MnO₃/BaZrO₃ composites. However, the refined grain microstructure by due to suppression of the La_{2/3}Ca_{1/3}MnO₃ growth by addition of BaZrO₃ makes the system to keep 3D continuum percolation behavior at room temperature.

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